Synthesis of Meso Bridgehead Dienes [3.3] Sigmatropic Rearrangements of 1,n-Divinyl Bicyclo [n,m,o] Alkanes

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Summary: [3.3]Sigmatropic rearrangement of l,n-divinyl bicyclo[n,m,o]alkanes results in formation of meso bridgehead dienes that contain a trans, trans-1,5-cycloalkadiene linkage.

Bridgehead dienes comprise such topologically interesting polycyclic structures as those illustrated in figures 1 and 2.1 These systems may be viewed as "locked" conformational isomers of trans, trans-cycloalkadienes. Meso derivative 2 contains an arrangement of two torsionally distorted bridgehead double bonds that optimizes the through space perturbation of the non-conjugated olefins. Selected derivatives of this ring system are now known.² We wished to develop a general synthesis for meso bridgehead dienes that would permit systematic exploration of the chemical and spectroscopic consequences of proximate relationships of bridgehead double bonds.



An earlier investigation of the thermal behavior of 3,6-bismethylene-1,7-octadiene³ revealed a plausible entry into the meso-bridgehead dienes (2). Our analysis is given below.



[3.3] Sigmatropic rearrangement of <u>cis-1</u>,2-divinylcycloalkanes can arise from two boat-like conformations.⁴ The endo conformation results in formation of a <u>cis,cis-1</u>,5-cyclooctadiene while rearrangement from the exo conformation produces a <u>trans,trans-1</u>,5-cyclooctadiene. In small and medium ring 1,5-cycloalkadienes the <u>cis,cis-</u>isomer is more stable;⁵ thermal rearrangement proceeds via the endo conformation.

If <u>cis</u>-divinyl groups occupy the bridgehead positions of bicyclo [n,m,o] alkanes, sigmatropic rearrangement will result in formation of a bicyclic <u>trans</u>, <u>trans</u>-1,5-cycloalkadiene, i.e. <u>meso</u> bridgehead diene 2. In the equation below the diene product is <u>trans</u>, <u>trans</u> in the B ring and <u>cis</u>, <u>cis</u> in the A ring. We anticipated that the <u>trans</u>, <u>trans</u>-1,5-diene component would be favored to reside in the largest ring. The overall position of equilibrium would dictate the success of the synthetic entry.



Synthesis of the required divinyl derivatives was accomplished by elaboration of the known bridgehead diesters.⁶ Thus LAH reduction followed by activated dimethylsulfoxide oxidation⁷ and bis Wittig olefination of the resulting dialdehydes affords good yields of divinyl compound. In several cases the divinyl product could not be isolated, rather spontaneous rearrangement to the bridgehead diene took place.⁸

The results are summarized in the Table. Several general trends can be made from these observations. In all cases, the equilibrium lies completely to the side of the bridgehead diene. Furthermore, a rough correlation exists between the strain energy of the bicyclic precursor and the rate of [3.3] signatropic rearrangement. For example, the lifetime of 1,4-divinylbicyclo [3.1.0]hexane (SE 33.5 kcal/mol) is short (minutes at 0° C) while 1,6-divinyl-bicyclo[4.1.0]heptane (SE 29.6 kcal/mol) has a half-life of 30 min at 25° C.



The proximate relationship of the two bridgehead double bonds is expected to manifest itself in the chemical behavior of these compounds. Preliminary results suggest this is so, for example, treatment of bridgehead diene $\underline{6}$ with Br₂ results in the uptake of one equivalent with formation of a single isomeric dibromide. The spectral properties of the adduct are consistent with the asymmetric dibromide (§) shown in the equation.⁹



We are continuing our investigation of the chemical and spectroscopic properties of these molecules as well as the synthesis of more highly strained members of this series.

Acknowledgment

We are grateful to the donors of the Petroleum Research Fund and to the National Science Foundation for support of this work. We also wish to thank Professor Philip Warner for providing us with information regarding his efforts in this area prior to publication.

References and Footnotes

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 - 4 ^LH NMR (250 MHz, CDCl₃) & 5.64 (m, 2H), 4.90 (m, 2H), 3.3 (d, J=3 Hz, 1H), 3.1-2.4 (m, 6H), 2.10 (m, contains d, J=13 Hz, 3H); ¹³C NMR (62.9 MHz, CDCl₃) & 144.4, 126.5, 117.7, 38.5, 34.5, 25.5; mass spectrum, m/e (EI, 70 eV), 146 (M+).
 - ⁵ ^LH NMR (80 MHz, CDCl₃) & 5.75 (m, 2H), 3.16 (d, J=12 Hz, 1H), 3.0-1.5 (m, 13H); ^{L3}C NMR (62.9 MHz, CDCl₃) & 147.7, 118.2, 36.8, 32.3, 27.4, 25.4; mass spectrum, m/e (EI, 70 eV) 148 (M+).
 - 6¹H NMR (250 MHz, CDCl₃) & 5.15 (m, 2H), 2.7-1.1 (m, 14H), ¹³C NMR (62.9 MHz, CDCl₃) & 144.3, 125.2, 98.6, 38.3, 29.6, 28.0, 25.0; mass spectrum m/e (CI) 149 (MH+).
 - ⁷ ¹ NMR (250 MHz, CDCl₃) & 4.63 (m, 2H), 2.90, 1.90 (AB, J=10 Hz, 2H), 2.60-2.00 (m, 10H); ¹³C NMR (62.9 MHz, CDCl₃) 155.4, 114.0, 41.5, 40.9, 37.7, 24.4; mass spectrum m/e (CI) 135 (MH+); UV (cyclohexane) 251 nm.
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(Received in USA 6 June 1983)